## Water Chemistry 11.10.95

Please answer to 6 questions from 7. The answers to the five first ones should be short (max. half page). 6 and 7 are essays. You are allowed to use Finnish or English.

- 1a) Liter of water contains 0,4 g of NaOH. What is the pH? ( $mw_{Na}$  is 22, 989)
- 1b) How do you make 500 ml 6 M nitric acid from 15 M solution?
- 2. The solubility of gas to liquid can be calculated by Henrys Law?a) What does it mean ? 3 p

b) You want improve the oxygen transfer efficiency of your aerator. What are the alternatives?

- 3. How do you analyze: (á 1/2 p)
  - a) taste and odor problems?
  - b) heavy metals?
  - c) radioactive substances?
  - d) humic compounds in aqueous solution?
  - e) dissolved oxygen?
  - f) pesticides form soil sample?

What kind of information is obtained by the following analyses?

g) conductivity

- h) pH
- i) BOD7
- j) hardness
- k) AOX
- d) alkalinity

- 4. A) What are the differences between municipal and industrial waste waters (3p)
  - B) Why water is an excellent solvent? (3p)
- 5. a) Explain what is the working principle of soaps and detergents (3p)

b) Inorganic carbon species in water (3p) Epäorgaanisen hiilen esiintymismuodot vedessä (3p)

- 6. What is the basic principle behind the drinking water guidelines given by Finnish, EC, USEPA or WHO authorities ?
- 6. Mihin sosiaali- ja terveysministeriön, EU:n, USEPA:n, WHOn jne. antamat talousveden laatuvaatimukset perustuvat (6p)
- 7. Pohjaveden laatuun vaikuttavat tekijät (6p)
- 7. Explain what factors affect the ground water quality?